## M.Sc. 3rd Semester Examination, 2018 CHEMISTRY <br> (Organic Chemistry) <br> Paper : CHEM 302C <br> Course ID : 31452

The figures in the margin indicate full marks.
Candidates are required to give their answers in their own words as far as practicable.

1. Answer any five:
(a) Write down the advantages of "Green chemistry".
(b) What are the instruments used to determine the average size and crystalline nature of the nano particles?
(c) How methoxyl group can be determined for the structure determination of an alkaloid chemistry?
(d) Fill in the blanks:
(i) Alkynes are $\qquad$ donor than Alkene.
(ii) $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{3}+\mathrm{PhBr}$ on oxidative adition form $\qquad$ .
(e) Indicate the isoprene units of the following compound:

(f) What is meant by supercritical fluid?
(g) Give an example of non-toxic metal and metal oxide nanoparticle, respectively.
2. Answer any four:
(a) State the factors that depends on the designing a green synthesis. Write two important uses of polymer nano particles.
(b) What do you mean by 'sol' and 'gel'? Mention basic principles of green chemistry.
(c) Name the appropriate green synthetic reagents for the following:
(i)

(ii)

(iii)

(iv)

(v)

(d) How acetyl-CoA converted into (s) $-\beta$-hydroxy- $\beta$-methylglutaryl-CoA by $\mathrm{CO}_{2}$ and biotin during biogenesis of monoterpines? Indicate the role of biotin.
(e) Write the products of the following:
(i)

(ii)

(iii)

(iv) Write a short note on polymer nanoparticle.
(f) How will you synthesize $\beta$-myrcene taking acrolein and diethylmalonate as a starting materials?
3. Answer any one:
(a) Describe the synthesis of papaverine. How will you prepare silica nanoparticle by sol-gel process (mentioning reagents, starting materials, reaction condition and mechanism)? 5+5=10
(b) Complete the following reaction sequences:
$5+2+2+1=10$


(iii)

(iv)

